Using a separating funnel

Please switch off water baths when finished to prevent them boiling dry.



Separating funnels are used for carrying out solvent extractions in organic chemistry. They can come in different shapes and volumes, make sure you use a suitable volume for your extraction and never fill the separating funnel more than two-thirds full.



When transferring liquids into the separating funnel you should make sure that the stopcock is closed and that the funnel is adequately supported, using a clamp or a ring.



When pouring liquids into the separating funnel you may wish to use a measuring cylinder or a longstemmed glass funnel to minimise any spillages



Separating the layers is usually the part most find difficult. The main reason for this is not knowing which is the aqueous and organic layers. The table below lists a number of extraction solvents along with their relative densities:

Solvent	Density (g/cm ³)
Petroleum ether	Typically ~ 0.640
Diethyl ether	0.714
Toluene	0.867
Ethyl acetate	0.900
Water	1.000
Dichloromethane	1.327

If your separation has gone well you should easily see two visible layers in the separating funnel. Here we have added colour to the aqueous and organic layers to show the separation more clearly, but for your separations there will be no colour.



NOTE: In both separating funnels, the red layer is the aqueous layer. But how can this be?

Explanation: Look at the table on the previous slide. In the left separating funnel, the aqueous layer is on the bottom, meaning the organic layer must be less dense than water. In the right separating funnel, the aqueous layer is on the top, meaning the organic layer must be more dense than water.

Video 1

Screen capture walkthrough, with annotations

In order to carry out an efficient extraction the organic and aqueous layers must be thoroughly mixed. Here is a series of pictures that are screen captures of an accompanying video. Note: CO₂ is evolved during extractions using aqueous carbonate or bicarbonate solutions!

Be aware of pressure build-up!

1. Insert the stopper





Ensure solid grip is used when holding the separating funnel. One option is to put the stem between your forefinger and middle finger

Stem of flask

3. Place one hand around the stopcock and keep the other over the stopper at all times



Ensure the stopper is firmly in the palm of your hand that you have a good grip at the top of the flask

To reiterate, ensure solid grip is used when holding the bottom of the separating funnel.

 Invert the funnel and immediately open the stopcock to release any pressure build up - when venting a funnel never point the stem towards your neighbours or yourself



Note: Whilst not shown to the left (for ease of demonstration), always try and keep two hands on the separating funnel. This is shown properly in a few slides

Tap open: allows venting

Always ensure you close the tap before turning it the right way round!

5. Close the stopcock. Continued over page



- 5. Close the stopcock and gently shake the mixture, release the pressure again by opening the stopcock
- 6. Repeat step 5 until no more vapour is expelled from the tap

Close tap

Gently shake the mixture

Release pressure by venting. <u>Note</u> tap is open.



- 7. Close the stopcock and return the separating funnel to the ring
- 8. Remove the stopper

Close tap



Return to retort ring



Remove stopper (essential for next step.



Video 2

Screen capture walkthrough, with annotations.

Here is a series of pictures that are screen captures of an accompanying video. Note: CO₂ is evolved during extractions using aqueous carbonate or bicarbonate solutions!

Be aware of pressure build-up!

It is now time to separate the layers into two different conical flasks, one for your aqueous layer and one for your organic layer (remember, colouring has been added here for emphasise – both your layers are likely to be colourless.

Note: This extraction is generic. You will need to determine whether you are extracting your aqueous layer with organic solvent, or vice versa. On this occasion, the red layer (bottom) is being extracted by the orange layer (top – called 'extraction solvent')

1a. Separate your two layers into two different conical flasks

Remember: The layer being extracted does not necessarily have to be the bottom layer. It could also be the top layer



1b. Separate your two layers into two different conical flasks

Put empty flask under separating funnel and collect extraction solvent

Nicely separated

- Return the layer being extracted (red here) to the separating funnel (perhaps use a long stemmed funnel to minimise spillages)
- 3. Add some more of your extraction solvent (orange), and repeat the shaking out process (see points 1-8 of video 1)

Return the layer being extracted

Pour in 'new' extraction solvent from measuring cylinder

Note the amount in measuring cylinder has decreased Back to similar situation as stage 1.

4. After shaking and venting (steps 1-8 of previous video), separate your two layers again, you can combine this portion of extraction solvent (orange) with the previous extraction solvent (orange)

Remove layer being extracted into its original flask Combine this portion of extraction solvent with previous portion

Nicely separated

- 5. Repeat steps 2, 3 and 4 once more
- You now have your two separate layers, make sure you know which one you need for further processing

Your extraction is finished. If the organic layer is the layer of interest (usually, but not exclusively), you should move onto drying the organic layer

